

Quantum Dots as Immunoassay Probes

Roger M. Leblanc



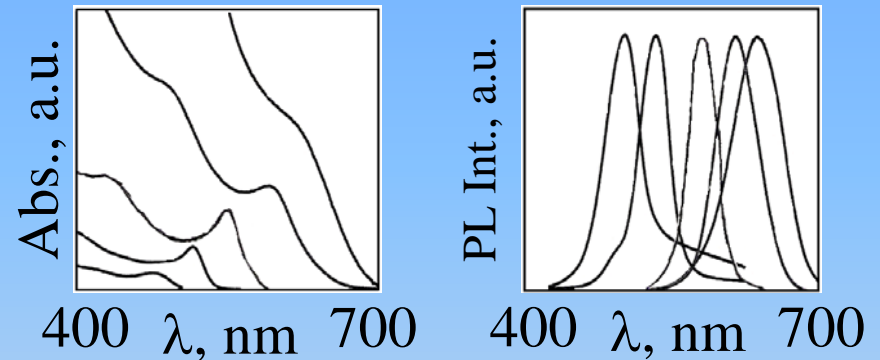
Introduction – quantum dots

Definition

- A material that confines electrons in 3D
- 1-10 nm length scale
(for most semiconductors)

CdSe QDs

(2.2, 2.6, 3.2, 4.3, and 5.5 nm)



Properties

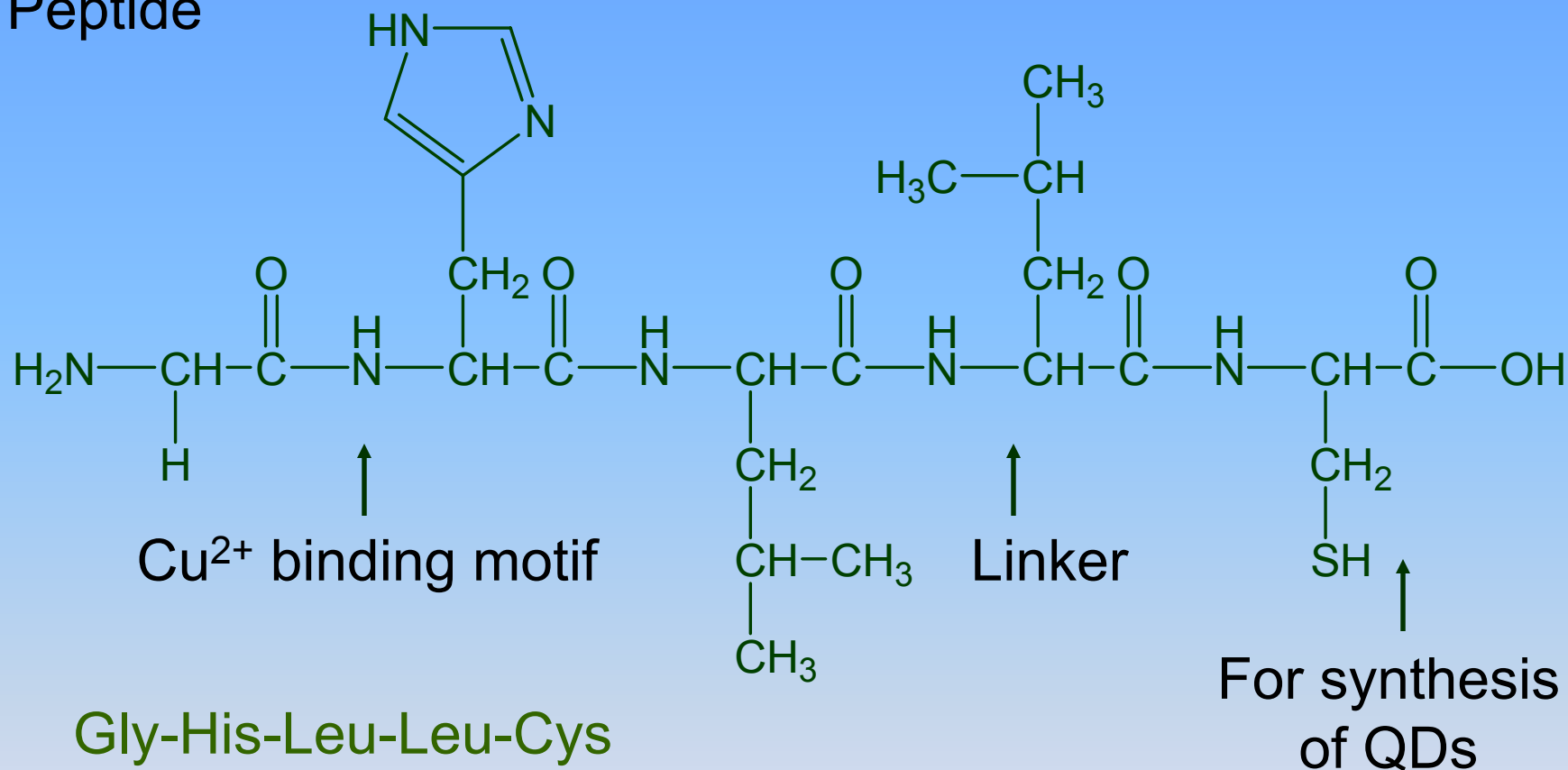
- Size dependent emission
- Narrow emission band: fwhm 35 nm
- High quantum yields: $\leq 85\%$
- Broad excitation spectra
- Chemical/photo stability



d = 1.9 2.4 4.1 5.2 nm
↑ ↑ ↑ ↑
 $\lambda_{\text{ex}} = 365 \text{ nm}$

CdS-Peptide QDs – Cu²⁺ Detection

Peptide



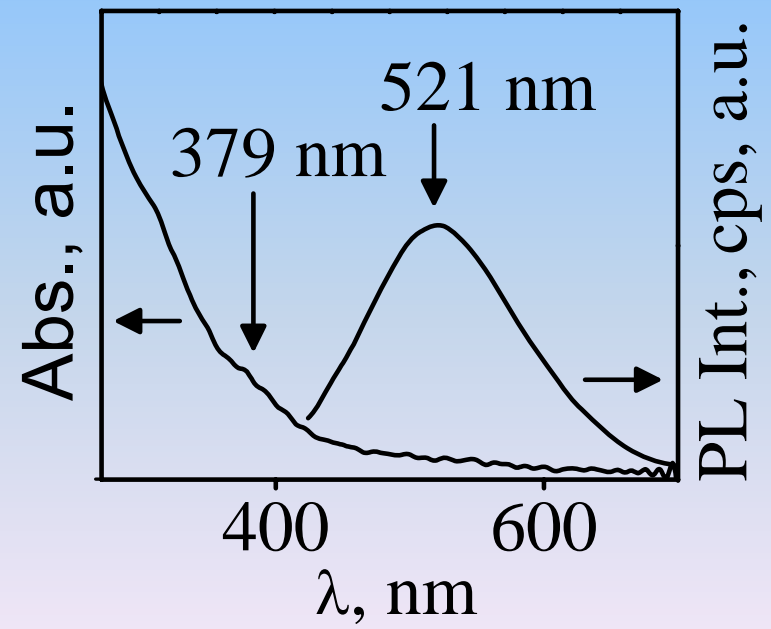
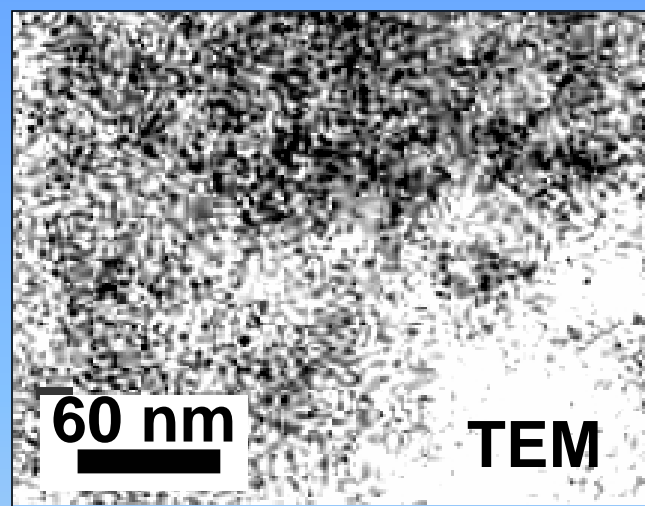
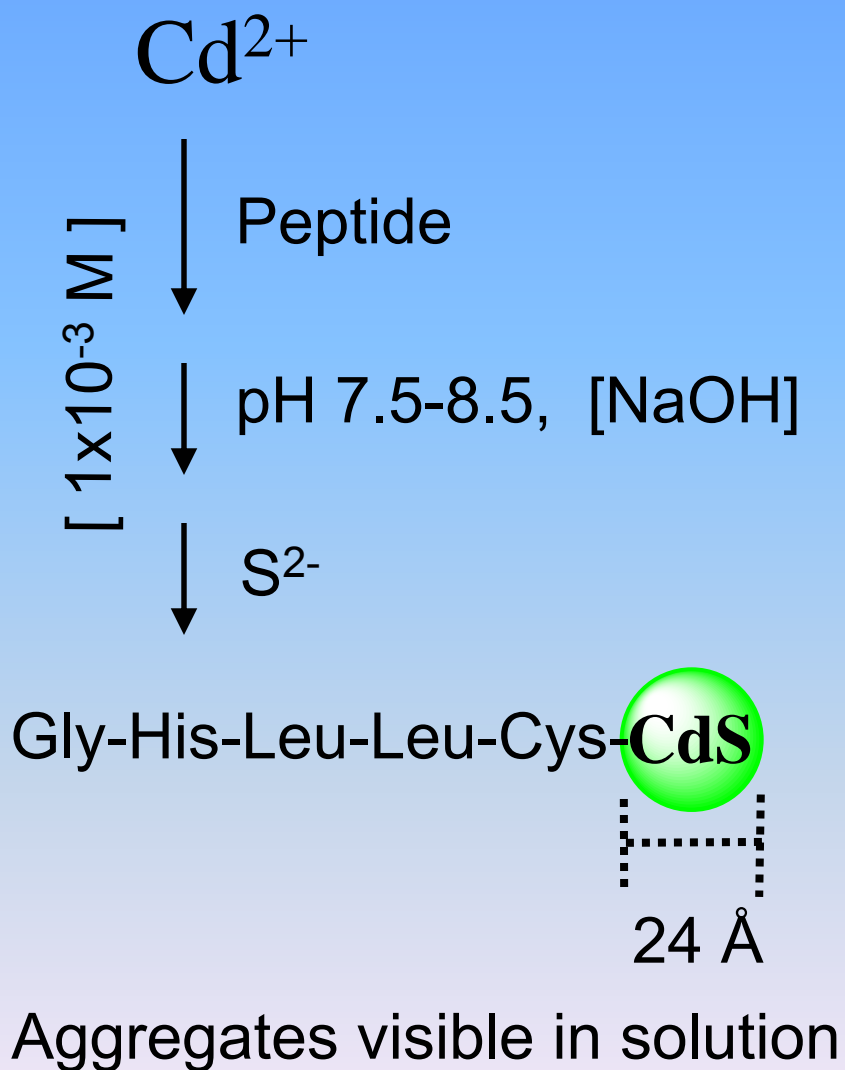
Gly-His-Leu-Leu-Cys

Fmoc solid phase synthesis, MW = 541.67

H₂O soluble, white powder, Yield = 79 %, Purity = 96 %

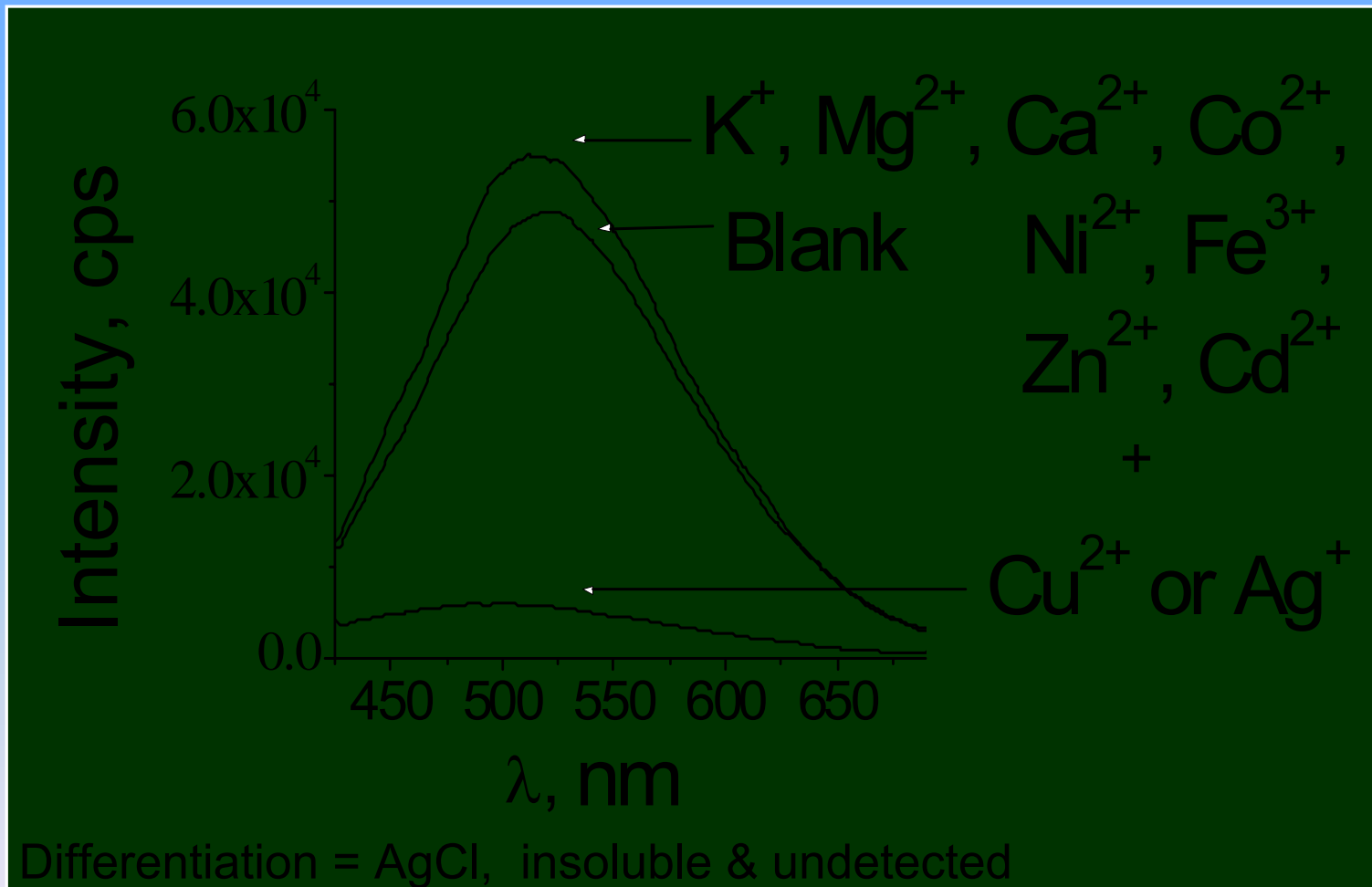
Gattás-Asfura, K. M. et al. Chem. Commun. 2003, 2684-2685.

CdS-Peptide QDs – Cu²⁺ Detection



CdS-Peptide QDs – Cu²⁺ Detection

Selectivity

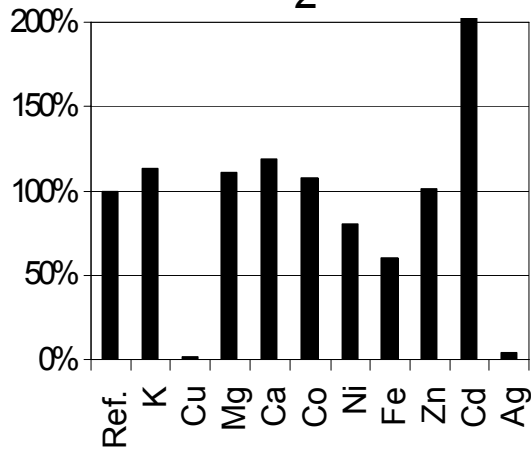


Gattás-Asfura, K. M. et al. Chem. Commun. 2003, 2684-2685.

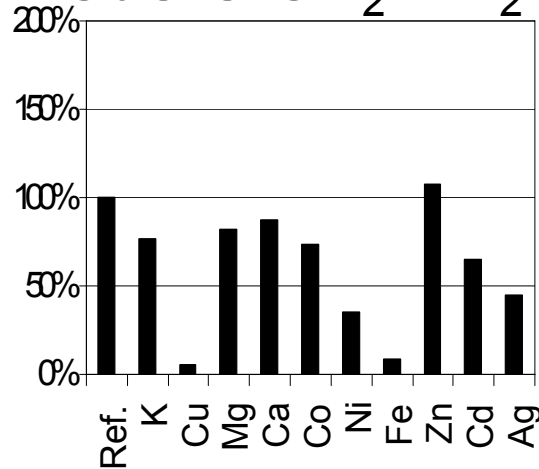
CdS-Peptide QDs – Cu²⁺ Detection

Selectivity, comparison

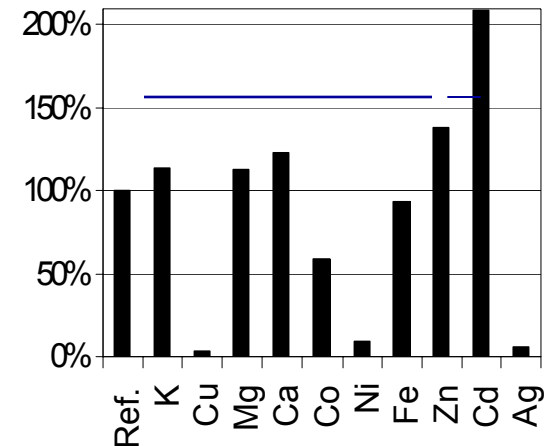
CdS-S-CH₂-COOH



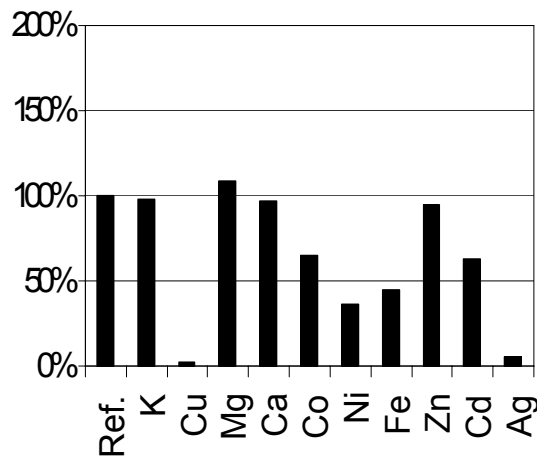
CdS-S-CH₂-NH₂



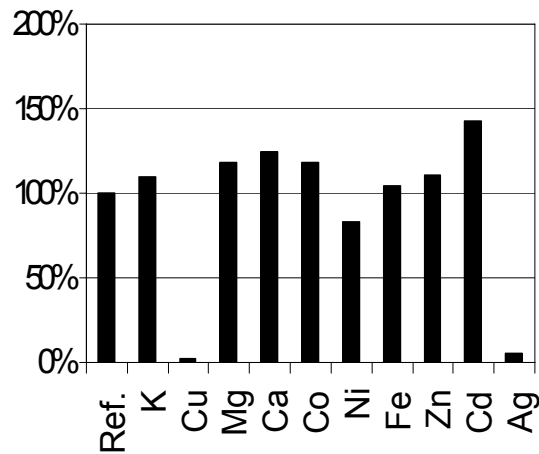
CdS-Cys



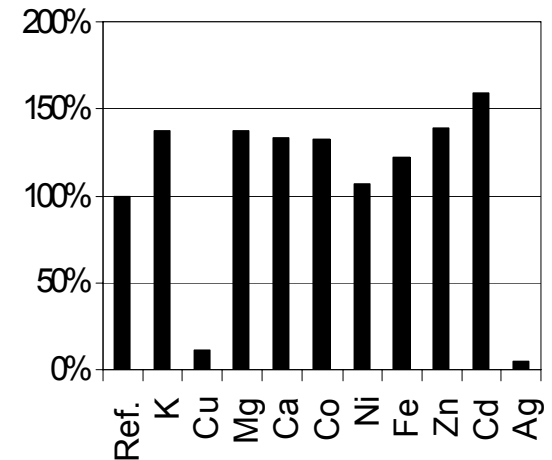
CdS-CGG



CdS-CLLGG



CdS-CLLHG



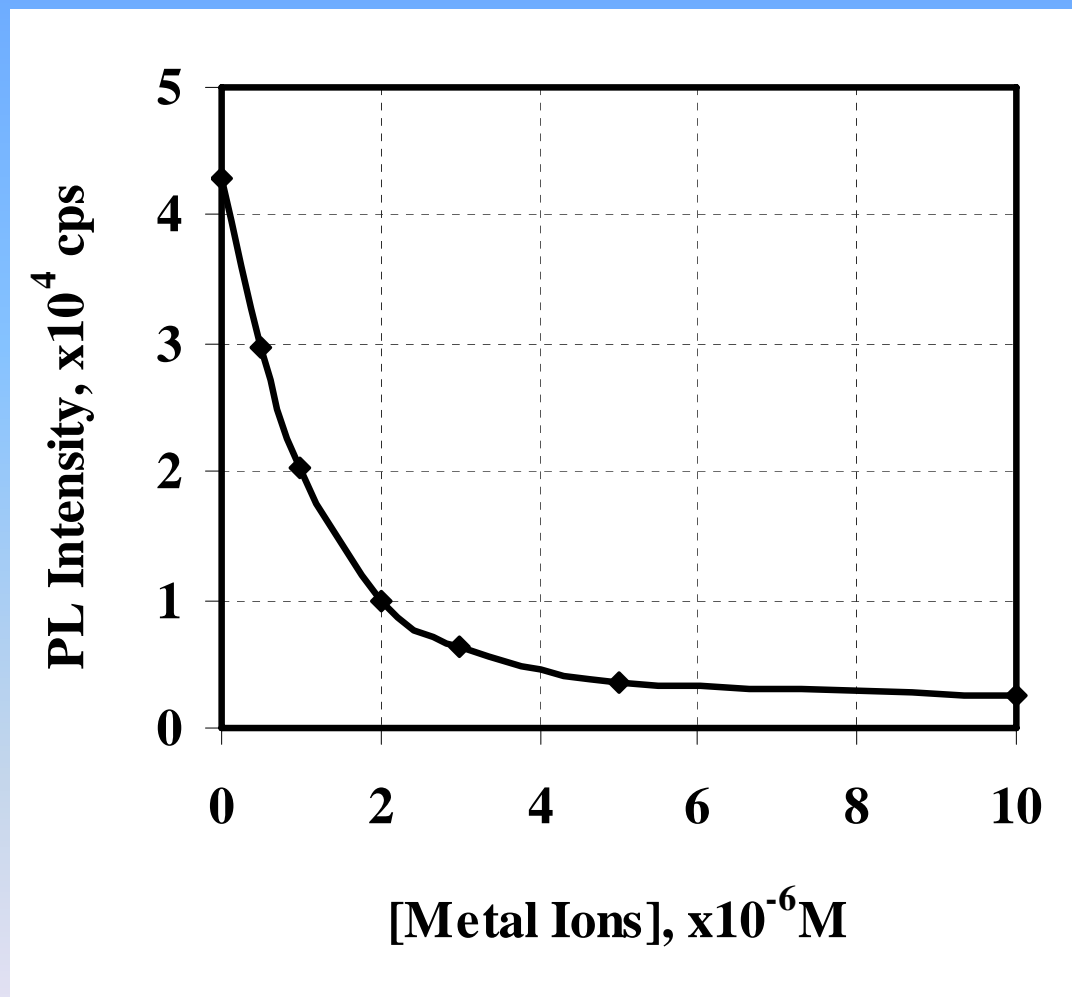
X = Metallic ions

Y = Relative PL intensity

CdS-Peptide QDs – Cu²⁺ Detection

Detected Levels

- Max. PL quenching at a Cu²⁺/QD molar ratio of 17:1 ($\epsilon_{\text{QD}} = 2 \times 10^4 \text{ M}^{-1}\text{cm}^{-1}$)
- $R^2 = 0.99$ exponential fit (nM detection capability)

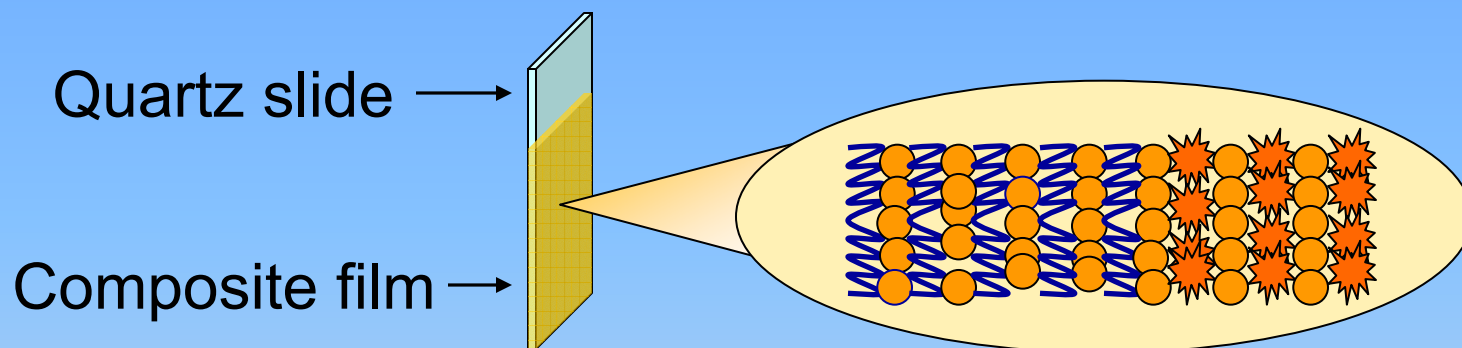


Conclusions


- CdS-Cys-Leu-Leu-His-Gly QDs detected Cu^{2+} and Ag^+ with high selectivity and sensitivity
- Leu-Leu and His-Gly residues played a critical role towards metallic ion selectivity
- Peptides facilitated fabrication of selective optical sensor by way of amino acid sequence design

CdSe QDs Composite Film – Paraoxon Detection

- ✓ Layer-by-Layer technique (electrostatic interaction)



 Chitosan

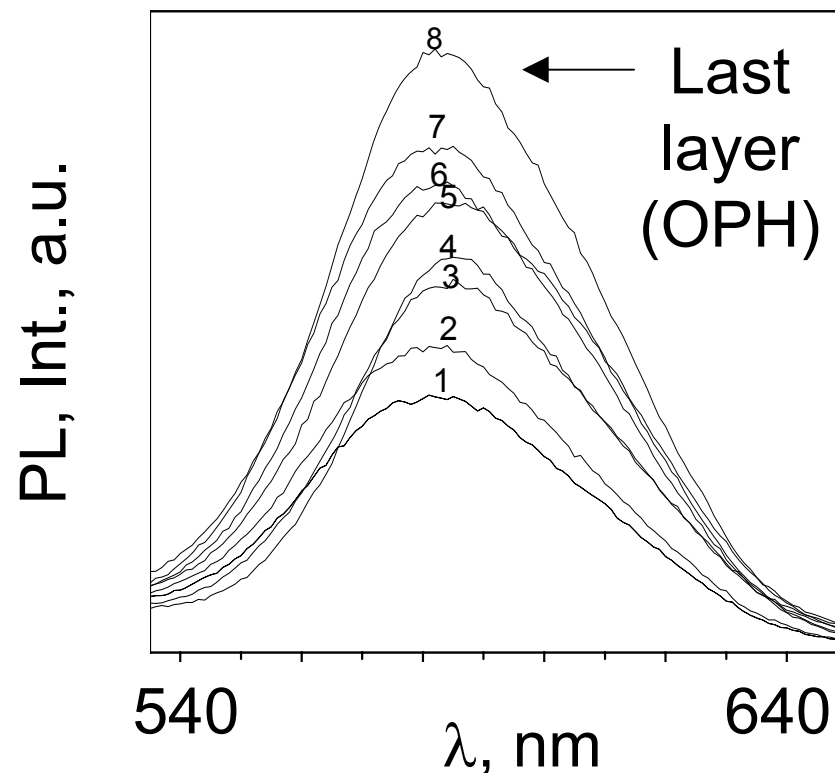
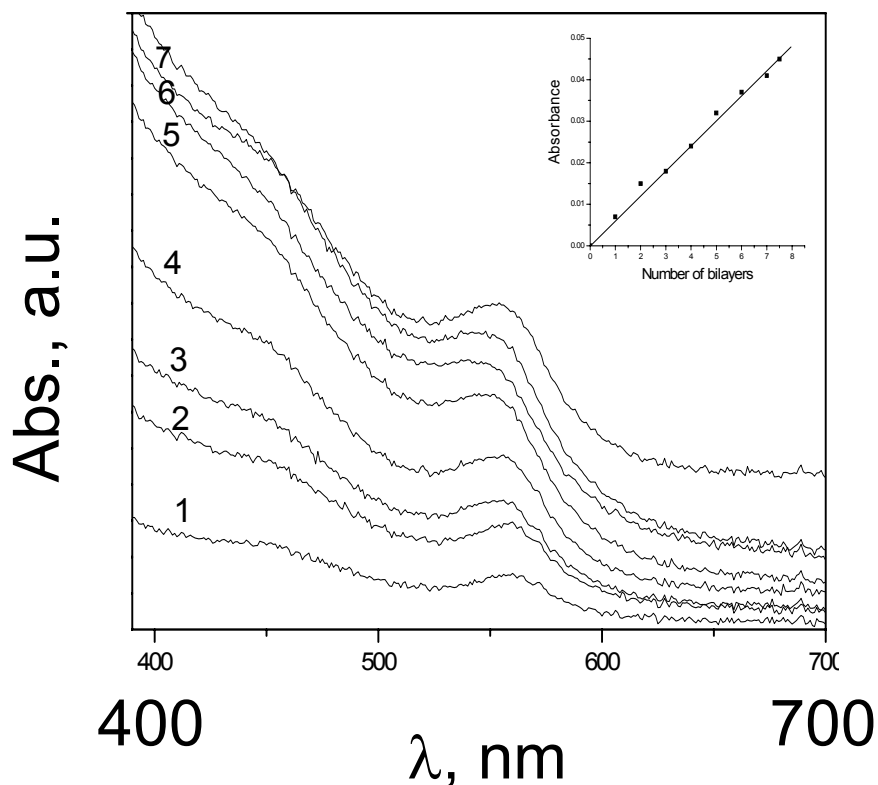
 CdSe-S-CH₂-COOH QDs

 Organophosphorus hydrolase (OPH)

Constantine, C. A.; Gattas-Asfura, K. M.; Mello, S. V.; Crespo, G.; Rastogi, V.; Cheng, T.-C.; DeFrank, J. J.; Leblanc, R. M. *Langmuir*; **2003**; 19(23); 9863-9867.

CdSe QDs Composite Film – Paraoxon Detection

Film growth



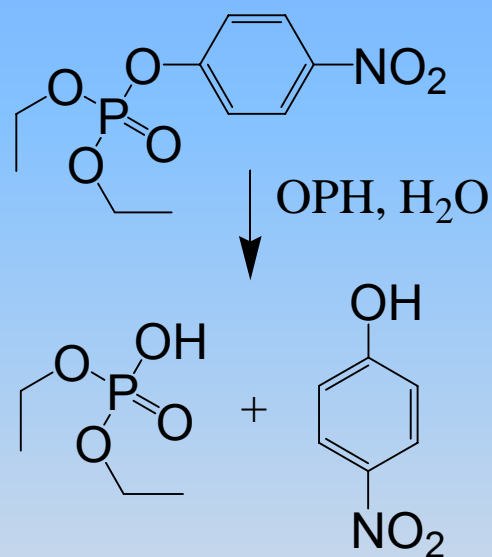
1-7 = # QDs layers

QD avg. size: 34 Å

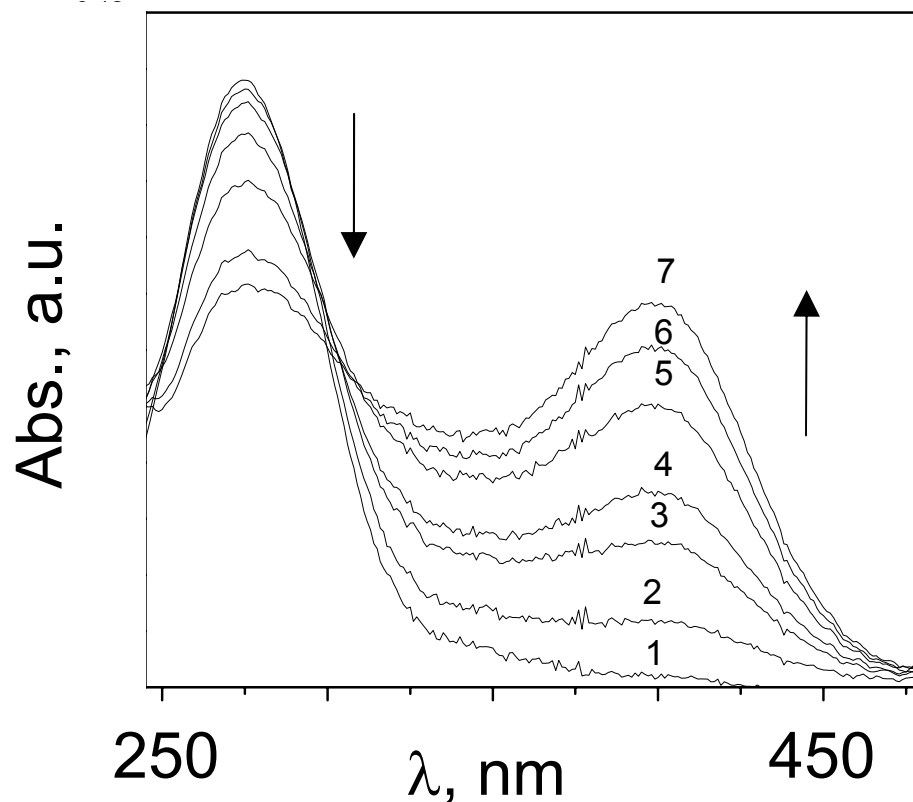
Constantine, C. A.; Gattas-Asfura, K. M.; Mello, S. V.; Crespo, G.; Rastogi, V.; Cheng, T.-C.; DeFrank, J. J.; Leblanc, R. M. *Langmuir*; **2003**; 19(23); 9863-9867.

CdSe QDs Composite Film – Paraoxon Detection

Response of film upon exposure to paraoxon



Paraoxon, 1×10^{-7} M

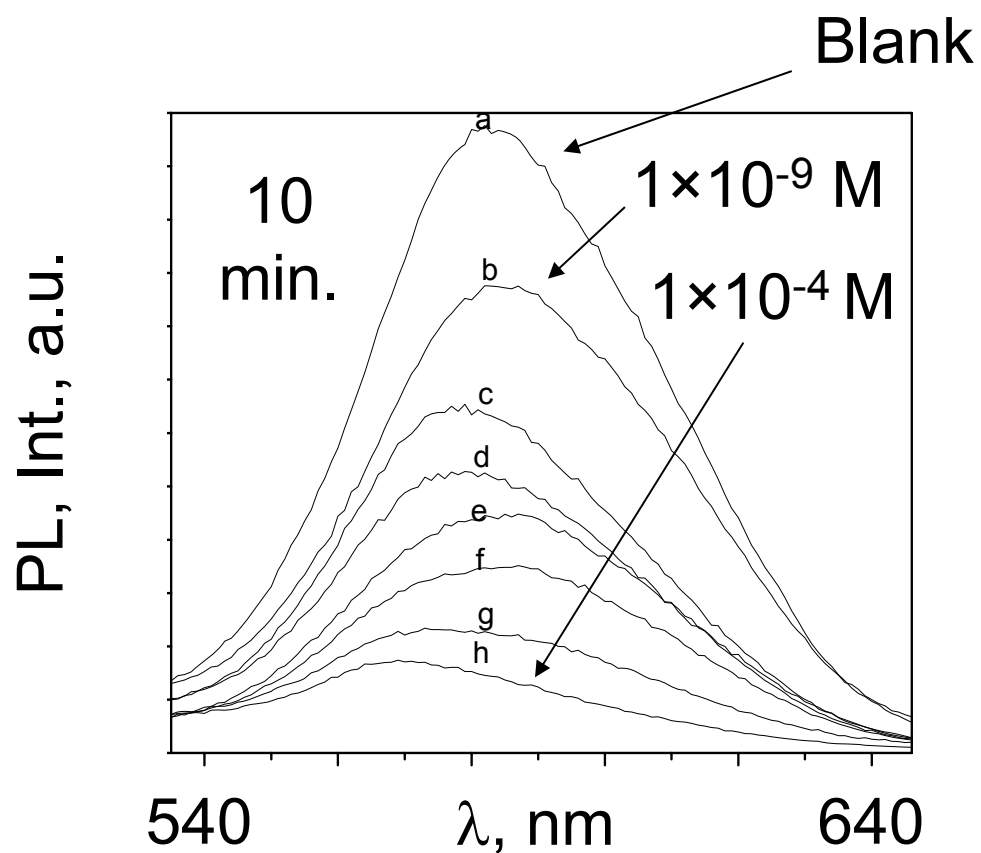
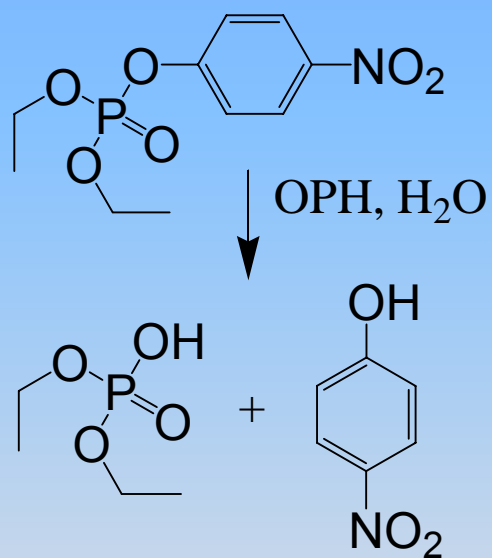


1-7 (bottom-top) = 0, 0.5, 1, 2, 5, 10, and 15 min.

Constantine, C. A.; Gattas-Asfura, K. M.; Mello, S. V.; Crespo, G.; Rastogi, V.; Cheng, T.-C.; DeFrank, J. J.; Leblanc, R. M. *Langmuir*; **2003**; *19*(23); 9863-9867.

CdSe QDs Composite Film – Paraoxon Detection

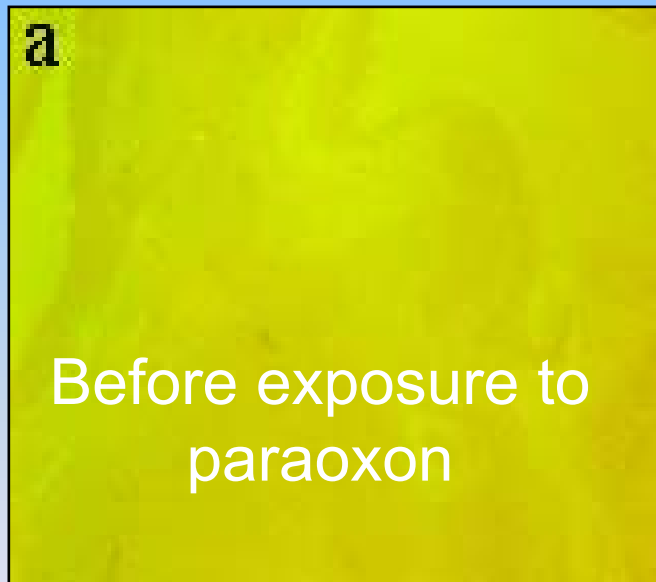
Response of film upon exposure to paraoxon



CdSe QDs Composite Film – Paraoxon Detection

Epifluorescence images of film

$895\mu\text{m} \times 793\mu\text{m}$

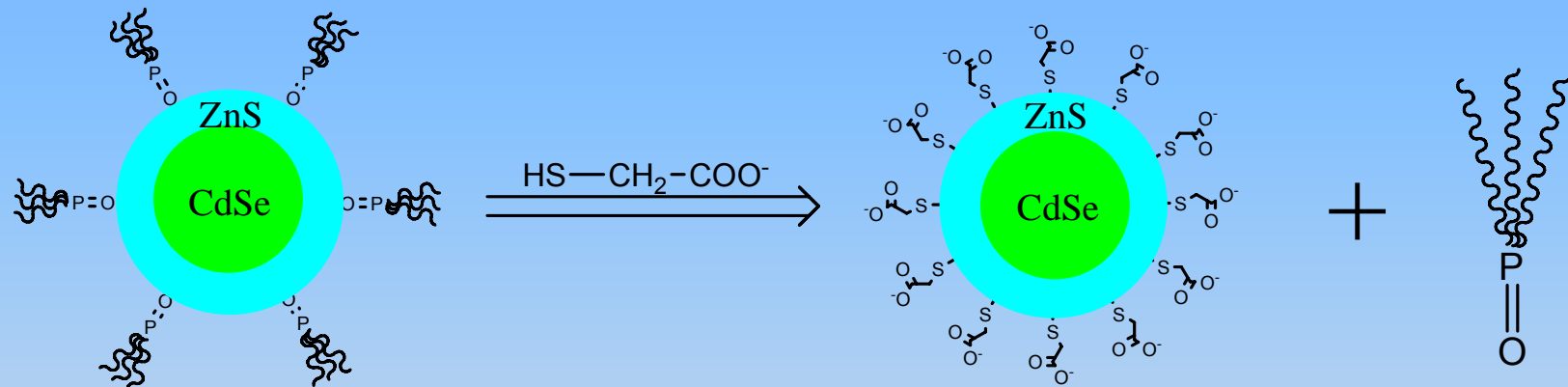


Conclusions

- QDs were successfully incorporated into composite films via the layer-by-layer technique and electrostatic interaction.
- The QDs/OPH ultra thin film selectively detected paraoxon at the nM levels.

Detection of paraoxon using QDs/OPH bioconjugate

Surface modification

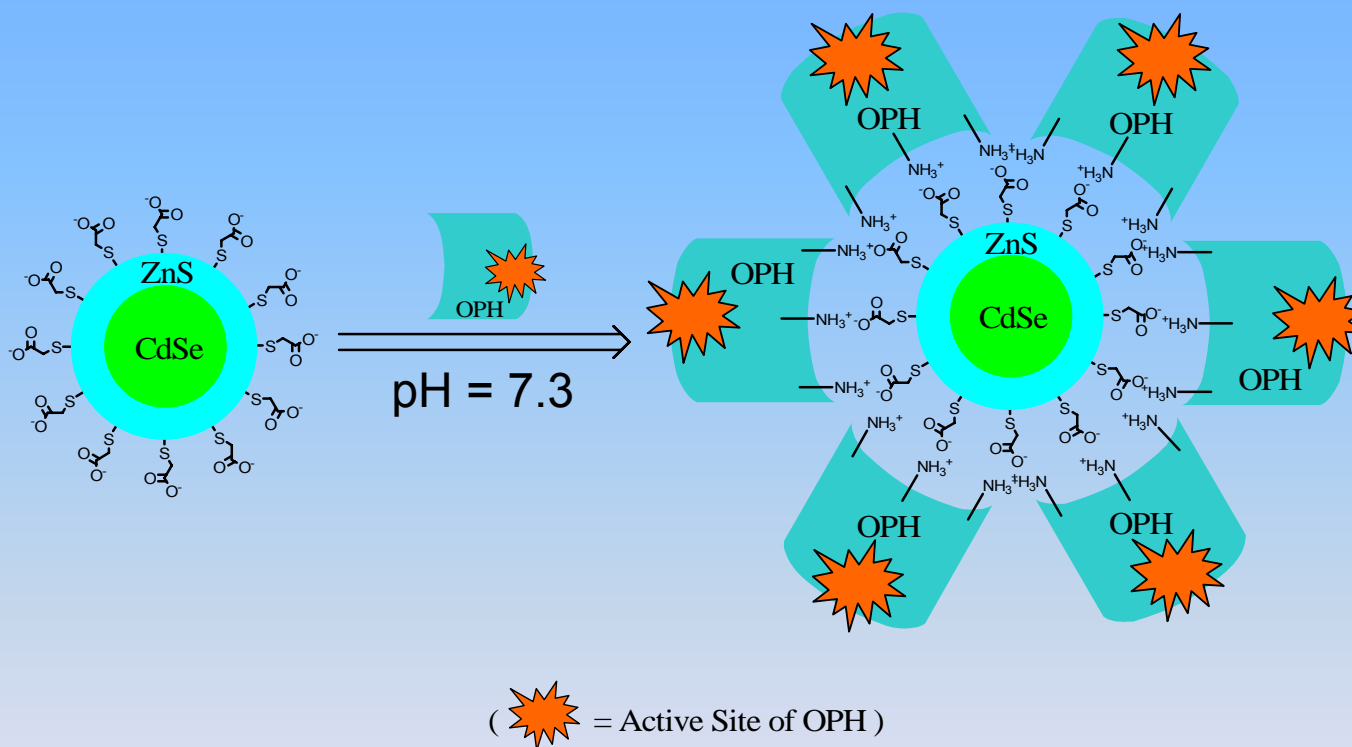


Hydrophobic

Hydrophilic

Detection of paraoxon using QDs/OPH bioconjugate

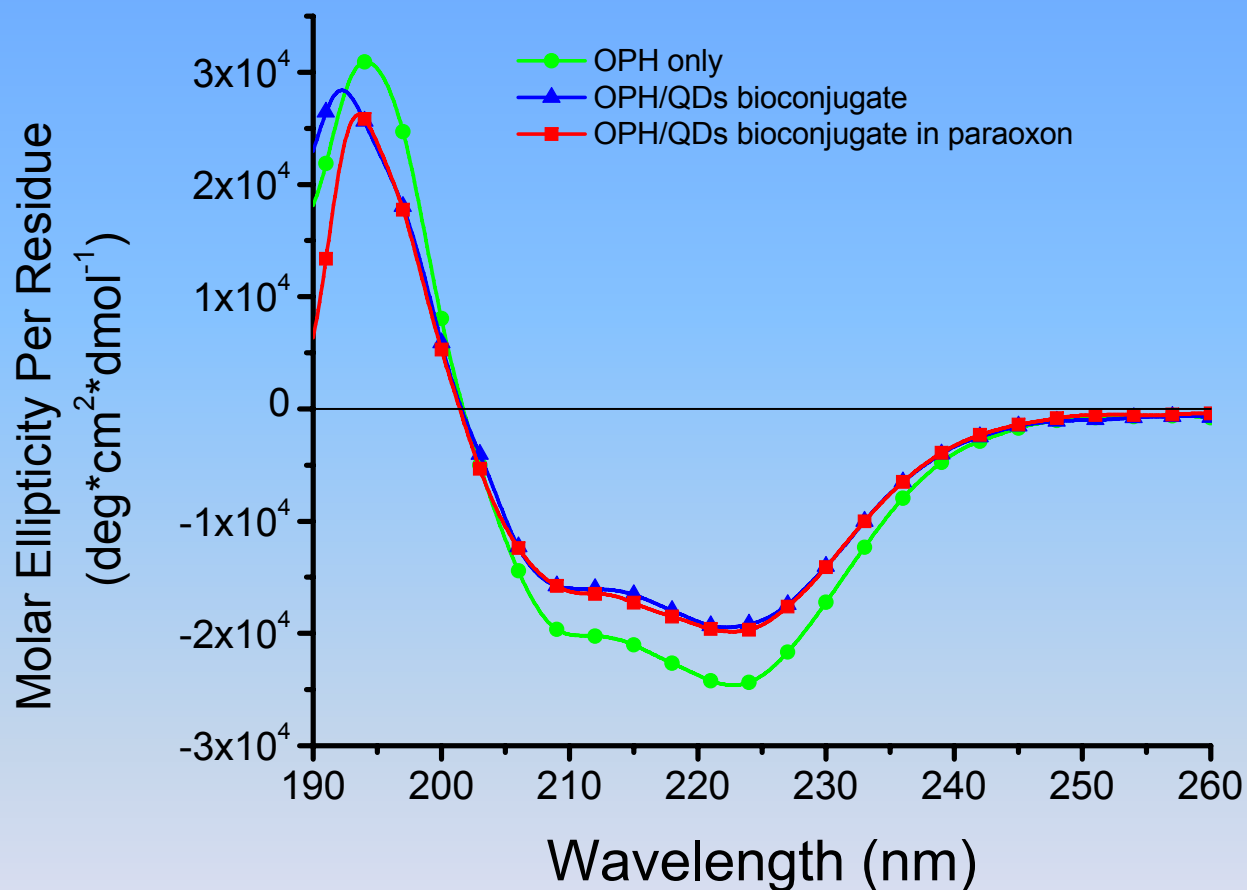
Bioconjugation



Isoelectric point of OPH : 7.6

Detection of paraoxon using QDs/OPH bioconjugate

Secondary structure analysis of bioconjugates



CD spectra of pure OPH and OPH/QDs bioconjugates with and without paraoxon. OPH concentration is 5×10^{-7} M in all cases. The OPH/QDs molar ratio of bioconjugates is 10. Paraoxon concentration is 10^{-6} M

Detection of paraoxon using QDs/OPH bioconjugate

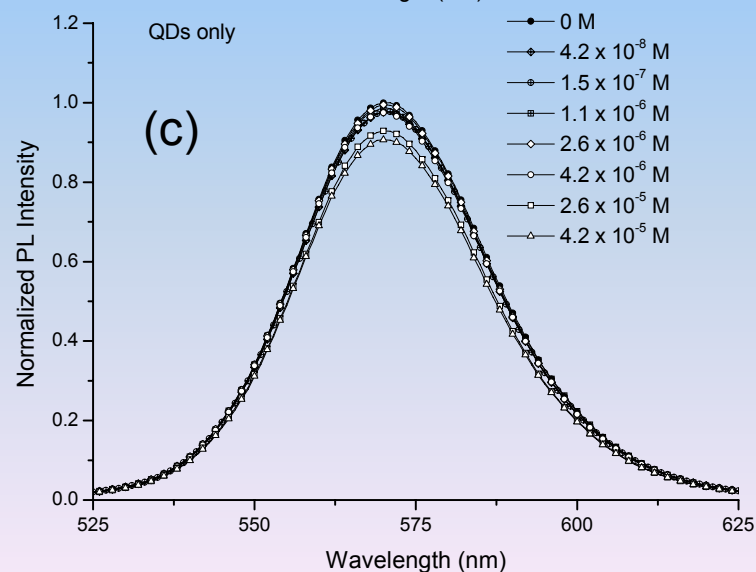
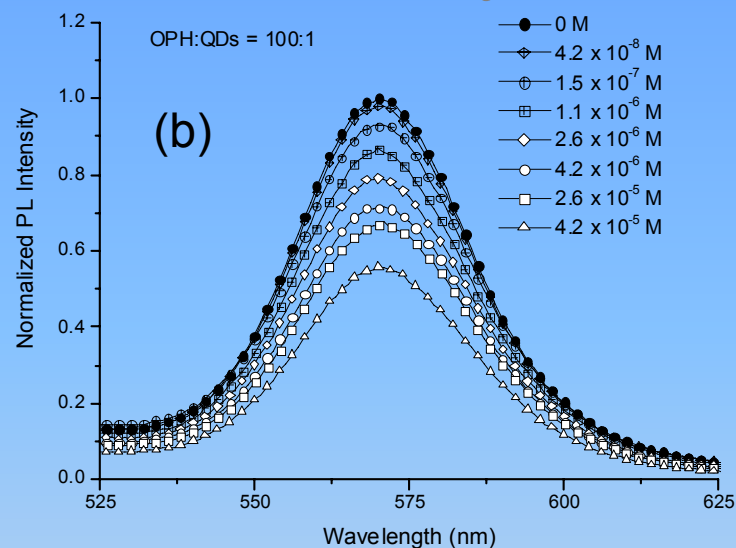
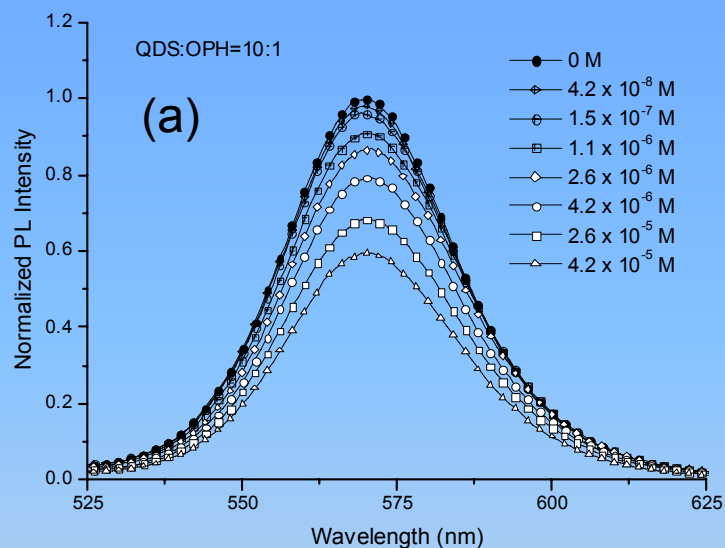
Secondary structure analysis of bioconjugates

	Percentage of Secondary Structure (%)					
	α -helix		β -strands		Turns (T)	Unordered (U)
	α_R	α_D	β_R	β_D		
OPH	40.03	25.53	2.40	4.43	13.67	15.53
OPH/QDs bioconjugates	36.10	21.43	3.37	4.10	13.10	20.07
OPH/QDs bioconjugates with paraoxon (10^{-6} M)	30.33	30.30	17.33	7.90	22.33	8.13

Secondary structure data of OPH, OPH/QDs bioconjugates and OPH/QDs bioconjugates in 10^{-6} M of paraoxon. The subscripts “R” and “D” represent “ordered” and “disordered”, respectively.

Detection of paraoxon using QDs/OPH bioconjugate

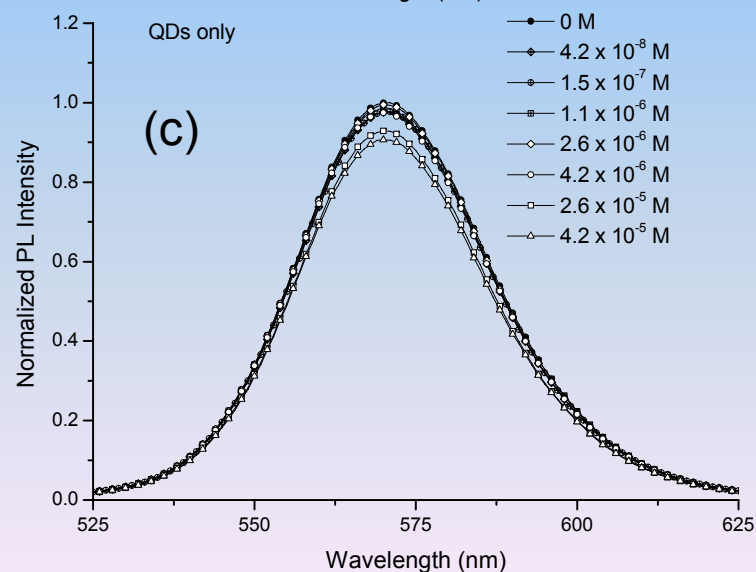
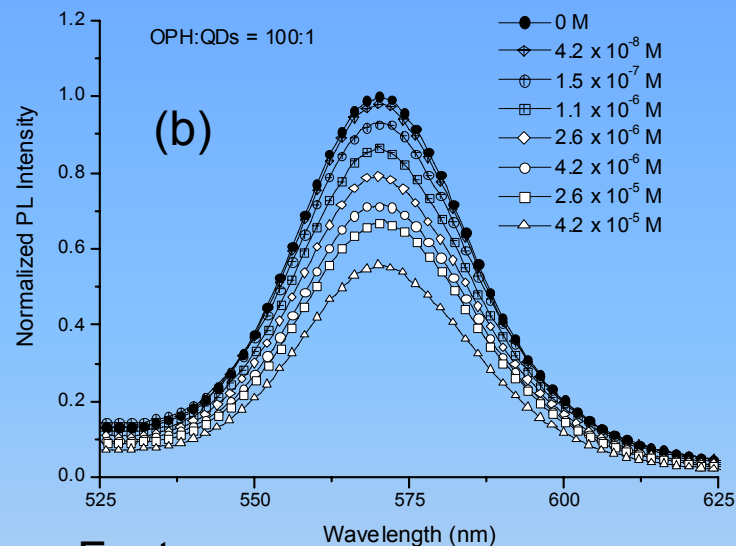
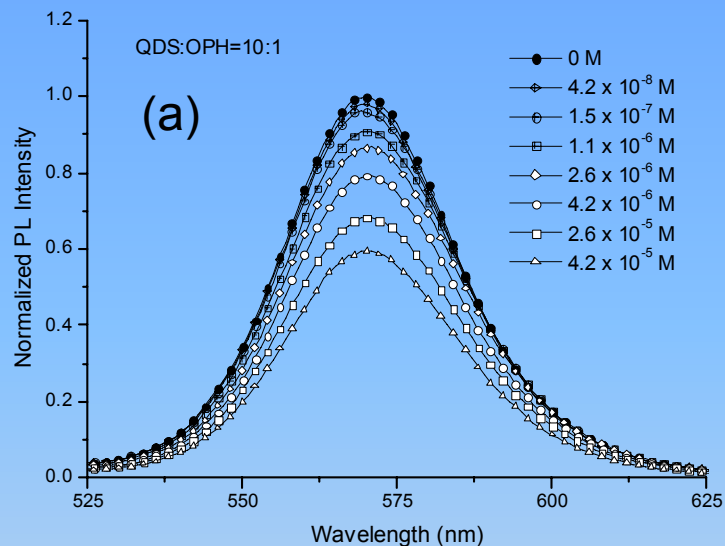
Fluorescence responses of bioassay



Photoluminescence spectra of (a) 10:1 and (b) 100:1 molar ratio OPH/QDs bioconjugates; (c) pure QDs in different concentrations of paraoxon. All samples were excited at 350 nm.

Detection of paraoxon using QDs/OPH bioconjugate

Fluorescence responses of bioassay

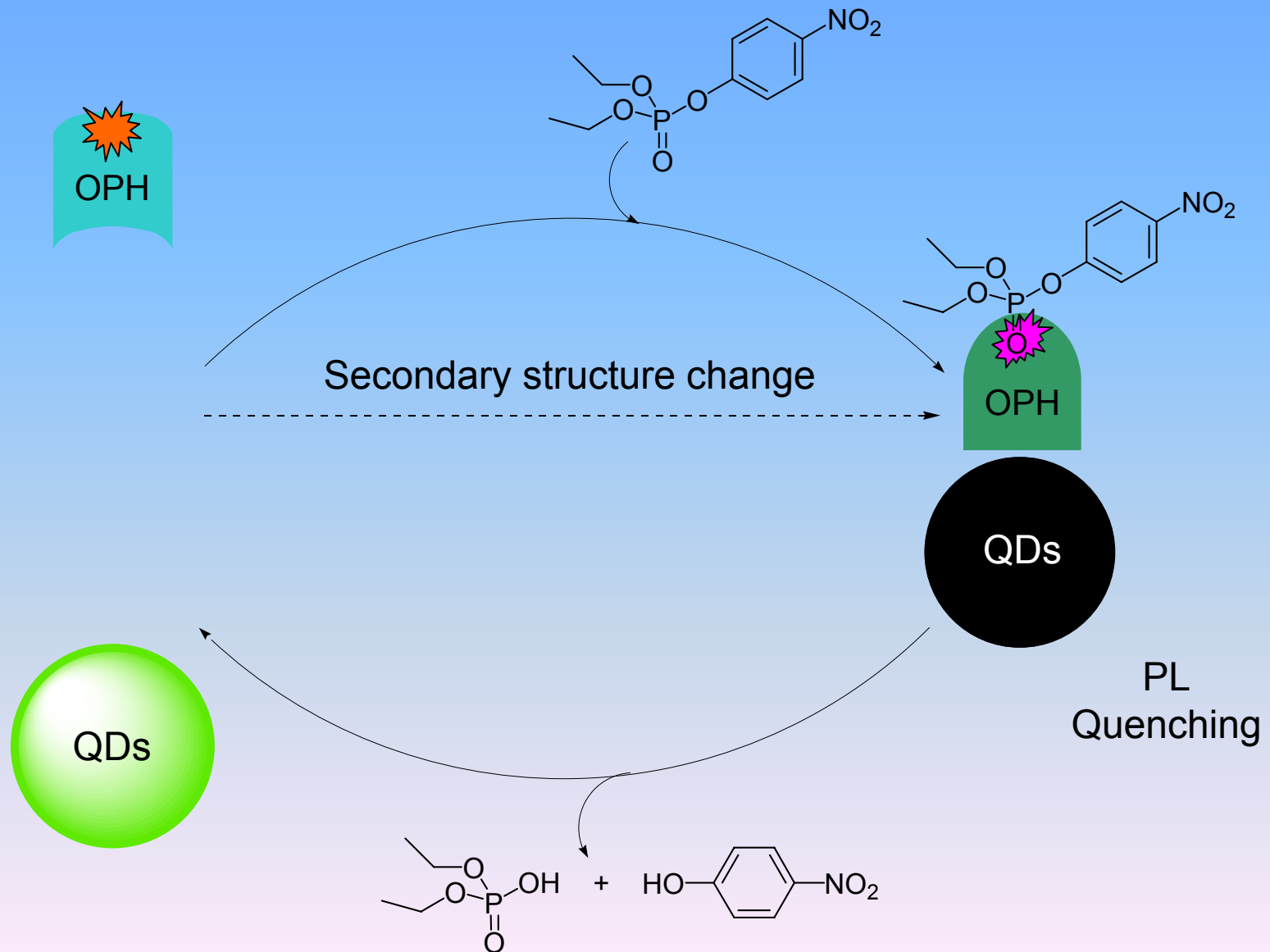


Facts:

1. The quenching of the PL is dependent to the concentration of paraoxon
2. The quenching of the PL has a maximum value in certain range of concentrations
3. The decrease of the PL intensity does not show a linear relationship to the [paraoxon]

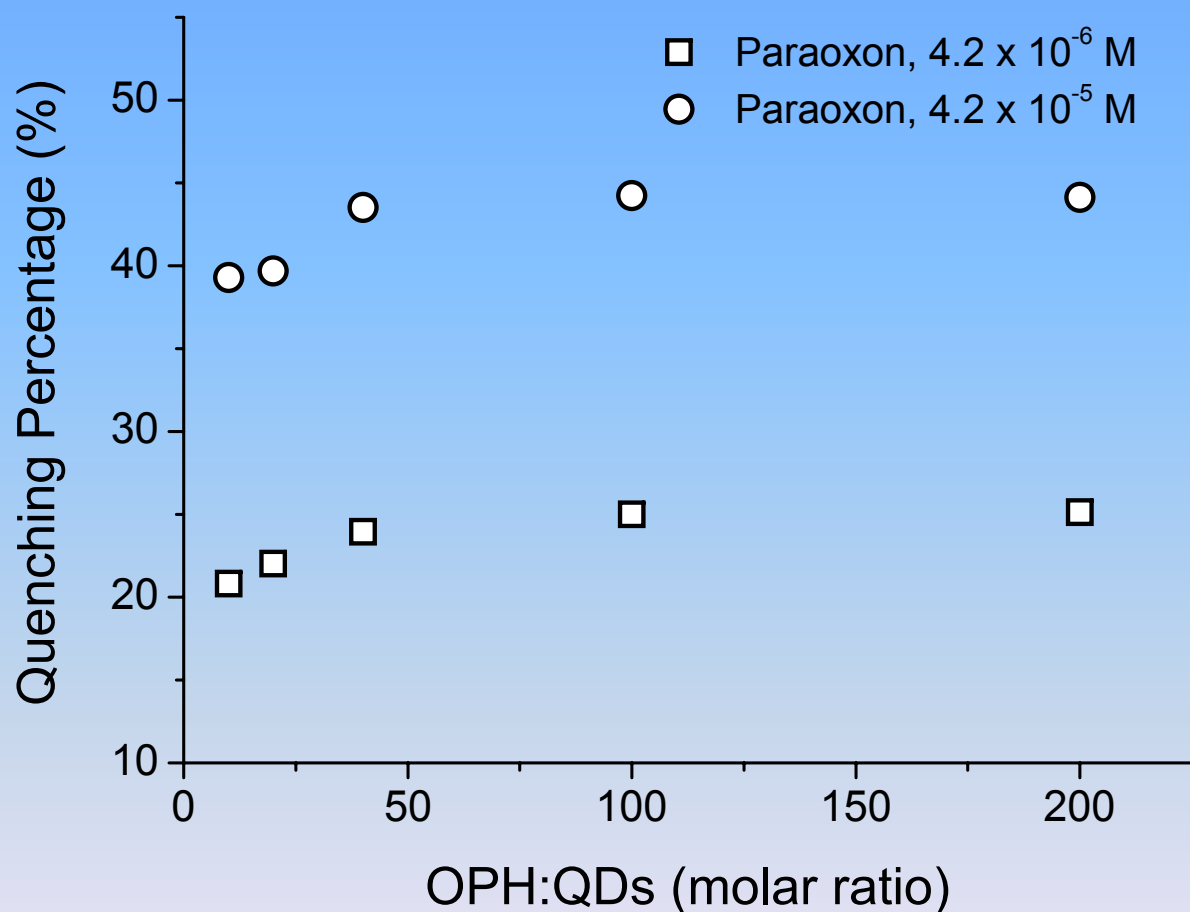
Detection of paraoxon using QDs/OPH bioconjugate

Proposed mechanism of quenching



Detection of paraoxon using QDs/OPH bioconjugate

Effect of OPH:QDs molar ratio on the sensitivity of the bioassay

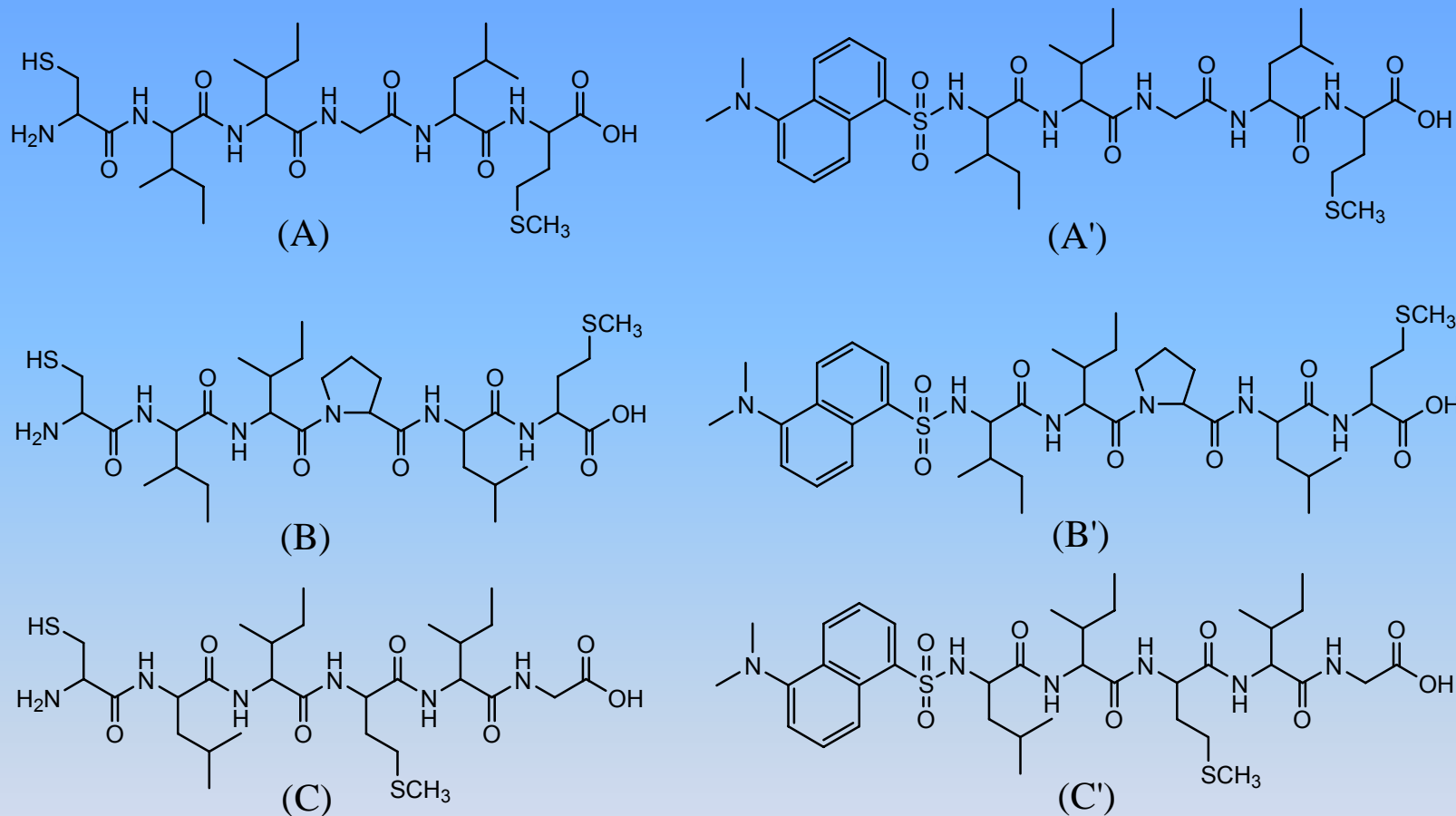


Concentration effect of OPH to the sensitivity of bioassay

Conclusions

- The OPH and CdSe(ZnS) QDs can form stable bioconjugate through electrostatic interaction
- CD spectrum indicate a secondary structure change of OPH in presence of paraoxon
- The intensity of photoluminescence of OPH/QDs bioconjugate was quenched in presence of paraoxon
- The secondary structure change of OPH was responsible for the observed PL quenching
- Increasing the molar ratio of OPH over QDs will not increase the sensitivity of OPH/QDs biosensor

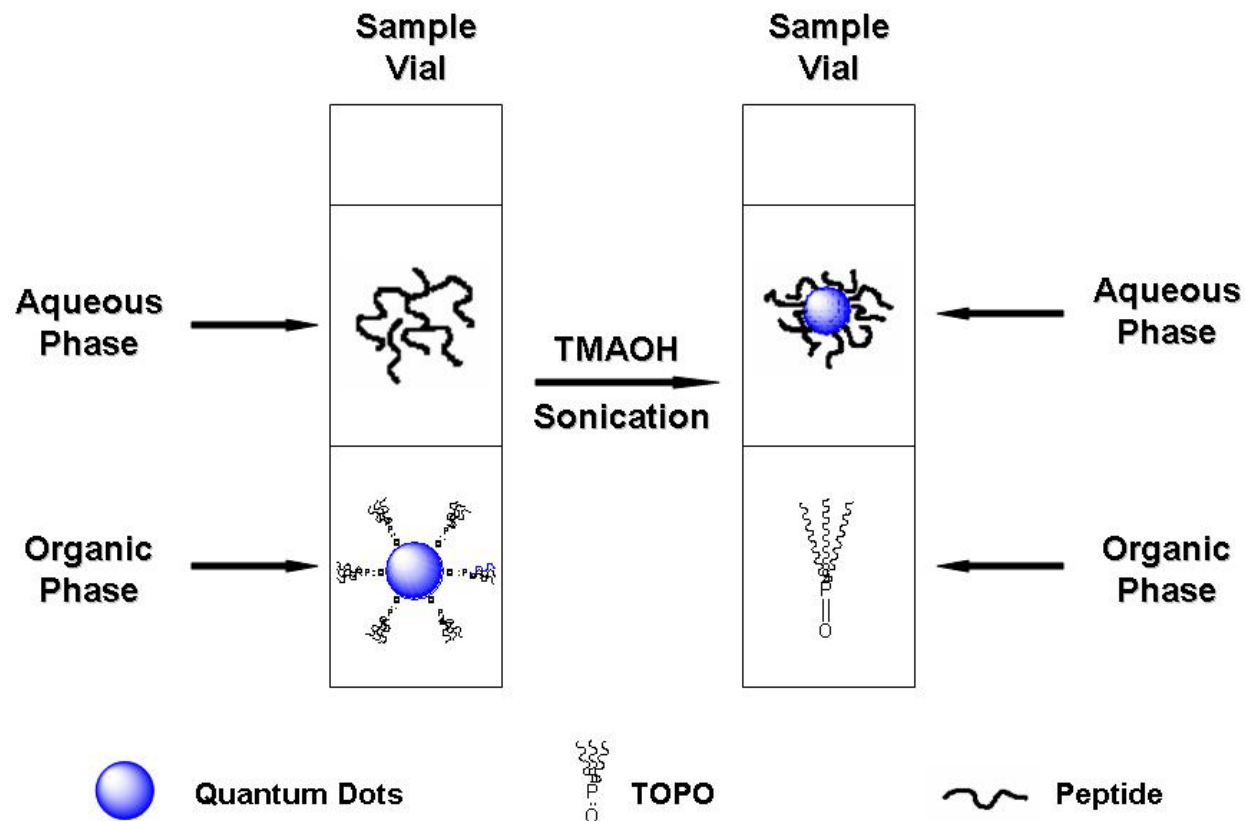
Imaging the formation of β -sheet using QDs/peptide conjugates



Structures of the synthetic peptides with cysteine linker and covalently bonded Dansyl group: (A) NH₂-Cys-Ile-Ile-Gly-Leu-Met-OH, (B) NH₂-Cys-Ile-Ile-Pro-Leu-Met-OH, (C) NH₂-Cys-Leu-Ile-Met-Ile-Gly-OH, (A') Dansyl-Ile-Ile-Gly-Leu-Met-OH, (B') Dansyl-Ile-Ile-Pro-Leu-Met-OH, and (C') Dansyl-Cys-Leu-Ile-Met-Ile-Gly-OH

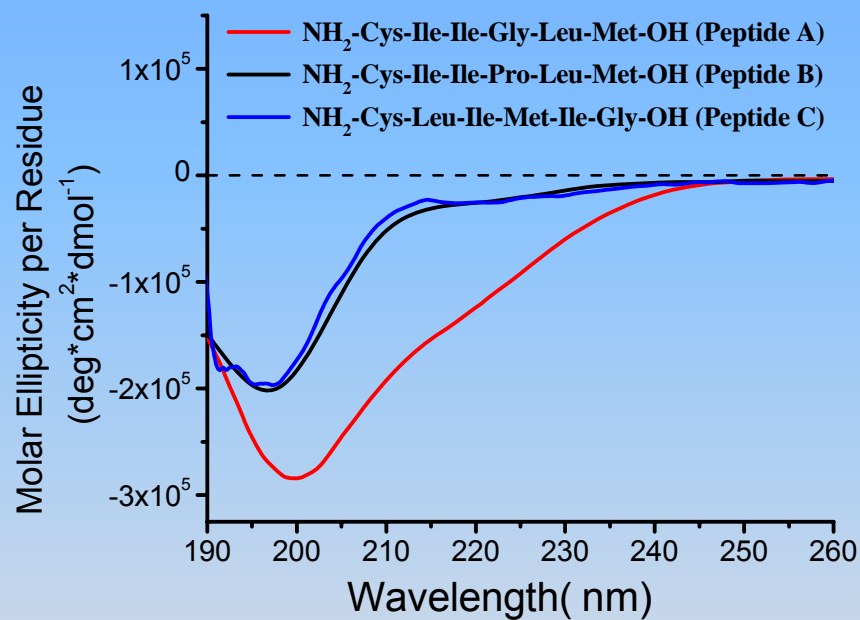
Imaging the formation of β -sheet using QDs/peptide conjugates

Schematic illustration of the surface coating of synthetic peptide on TOPO-coated CdSe/ZnS

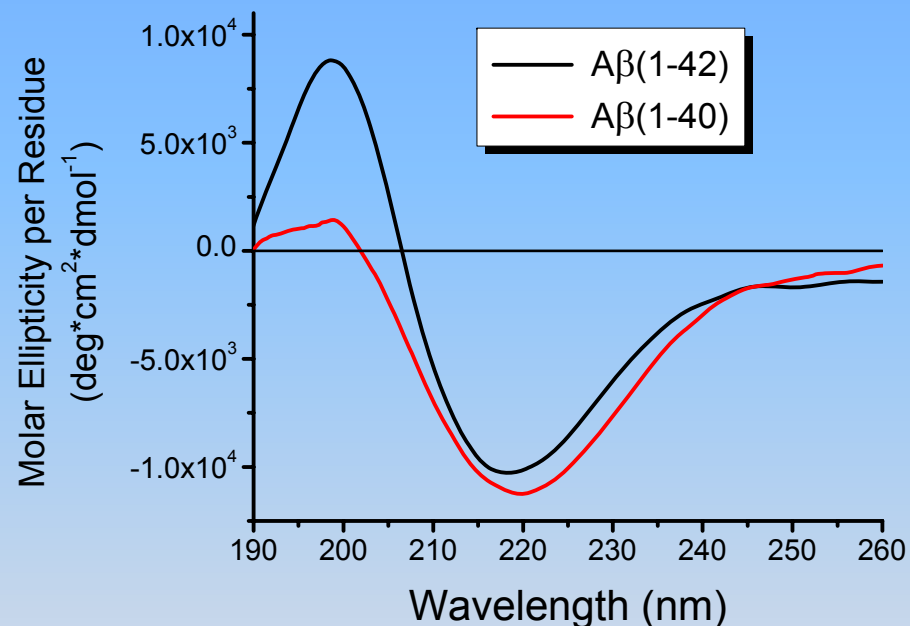


Imaging the formation of β -sheet using QDs/peptide conjugates

Steric configuration analysis by Circular Dichroism spectroscopy



(a)



(b)

Circular Dichroism spectra of (a) three model peptides and (b) full length peptides.

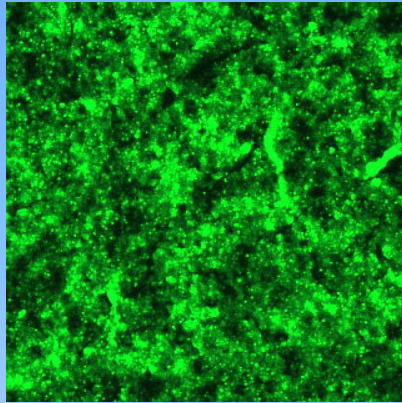
Imaging the formation of β -sheet using QDs/peptide conjugates

Secondary structural data of A β (1-40), and A β (1-42) peptides. The subscripts "R" and "D" represent "ordered" and "disordered", respectively.

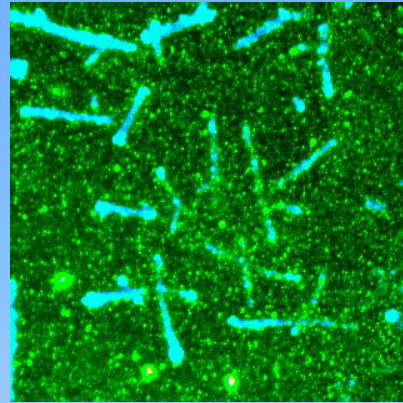
	Percentage of Secondary Structure					
	α -helix		β -strands		Turns (T)	Unordered (U)
	α_R	α_D	β_R	β_D		
A β (1-40)	12.5	11.3	18.5	8.5	19.0	30.2
A β (1-42)	11.8	9.5	22.6	14.5	14.0	27.5

Imaging the formation of β -sheet using QDs/peptide conjugates

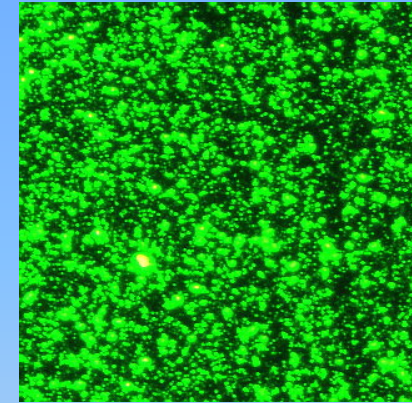
Epifluorescence micrographs of 31-35 model peptides and controlled peptides



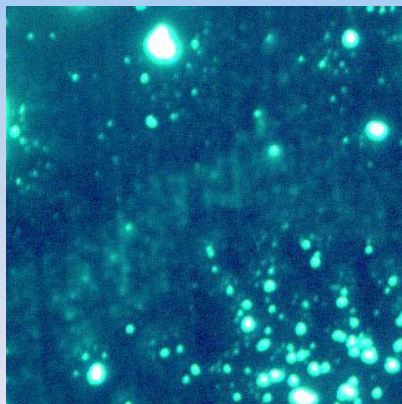
QDs-CIIGLM-OH



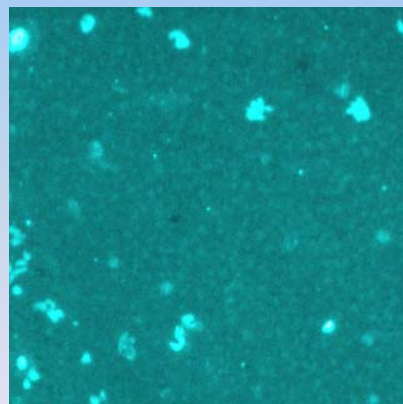
QDs-CIPLM-OH



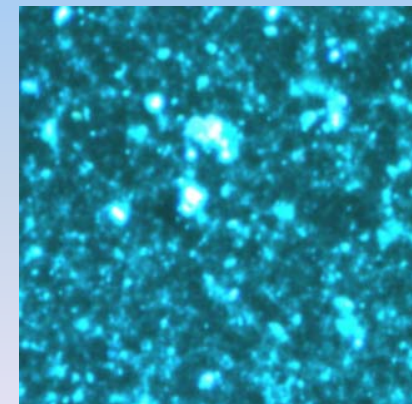
QDs-CLIMIG-OH



Dansyl-CIIGLM-OH



Dansyl-CIPLM-OH

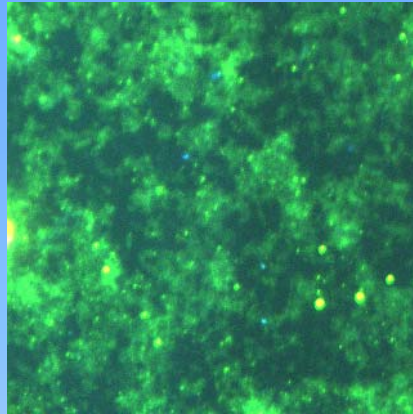


Dansyl-CLIMIG-OH

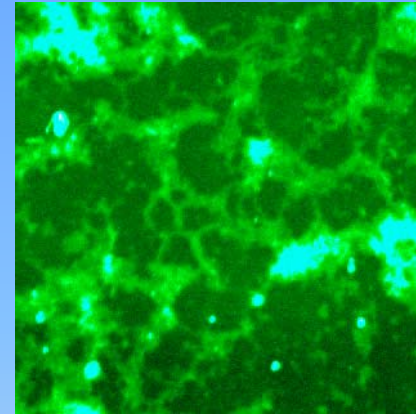
Ji X., D,Naistat, C. Li, J. Orbulescu and R.M. Leblanc, Colloids and Surfaces B: Biointerfaces **2006**, in press.

Imaging the formation of β -sheet using QDs/peptide conjugates

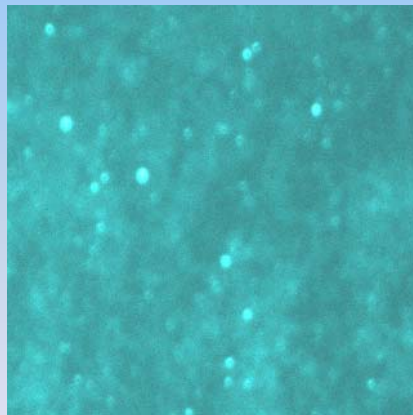
Epifluorescence micrographs of full length peptides



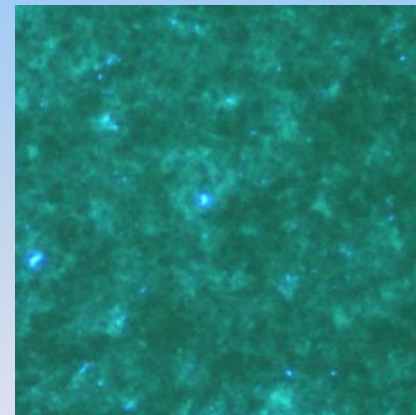
A β (1-40) / QDs
mixture



A β (1-42) / QDs
mixture



A β (1-40) / Dansyl
acid mixture



A β (1-42) / Dansyl
acid mixture

Conclusions

- The model 31-35 A β peptides with cystein linker and Dansyl group were designed and synthesized
- A phase transfer approach was applied to conjugate the peptides onto the surface of the QDs
- The steric configuration study by CD spectroscopy revealed that the main secondary structural component of the aggregation is β -strands
- Epifluorescence microscopic research showed that QDs luminescent label approach showed a much higher intensity and better contrast for imaging than organic dye



Group Members

PhD Graduate Students:

- Jiayin Zheng (March 2006)
- Kerim M. Gattás-Asfura (December 2005)
- Xiaojun Ji (December 2005)
- David Naistat (December 2005)
- Changqing Li (June 2005)
- Robert Triulzi
- Jianmin Xu
- Liang Zhao

Sr. Research Associate:

- Jhony Orbulescu

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